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AMENDMENTS TO THE CLAIMS

This listing of claims replaces all prior versions of listing of claims, and listing of claims in the application.

Listing of Claims

(Currently Amended) A liquid ionic compound comprising a cation which is a
complex of a neutral ligand selected from the group consisting of organic substituted and
unsubstituted alkyl amines and crown ethers with a metal ion selected from the group
consisting of Na⁺, K⁺, Li⁺, Ca²⁺, Ag⁺, Zn²⁺, Cu²⁺, Cd²⁺, Ni²⁺, Hg²⁺, Co³⁺ and Fe³⁺ and an
anion which is a conjugate anion of the metal ion, wherein said anion comprises sulfur or

 (Previously Presented) An ionic compound according to claim 1 which is a liquid below 100°C.

phosphorous said cation and said anion comprising a substantially new liquid.

- 3. (Previously Presented) An ionic compound according to claim 2 which is a liquid at room temperature.
- (Previously Presented) An ionic compound according to claim 1 which is electrically conductive in the absence of a solvent.
- (Previously Presented) An ionic liquid according to claim 1 which is hydrophobic.
- (Previously Presented) An ionic compound according to claim 1 wherein said neutral organic ligand is a crown ether.
- (Cancelled)

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- (Previously Presented) An ionic compound according to claim 1 wherein said conjugate anion is bis(trifluoromethane)sulfonimide, boron trifluoride, nitrate, sulfate, phosphate, hexafluorophosphate and dicyanamide.
- 9. (Currently Amended) A method for forming a substantially neat ionic liquid ionic compound, the method comprising mixing a neutral ligand selected from the group consisting of organic substituted and unsubstituted alkyl amines and crown ethers with a metal ion selected from the group consisting of Na⁺, K, Li⁺, Ca²⁺, Ag⁺, Zn²⁺, Cu²⁺, Cd²⁺, Ni²⁺, Hg²⁺, Co³⁺ and Fe³⁺ and with the salt of a metal cation and its conjugate anion at room temperature.
- (Original) A method according to claim 9 wherein said neutral organic ligand is a crown ether.
- 11. (Original) A method according to claim 10 wherein the metal cation is selected from the group consisting of sodium, potassium, lithium and calcium.
- 12. (Cancelled)
- 13. (Original) A method according to claim 12 wherein said metal cation is selected from the group consisting of silver, zinc, copper, cadmium, nickel, mercury and iron.
- 14. (Previously Presented) A method according to claim 9 wherein said conjugate anion is bis(trifluoromethane)sulfonimide, boron trifluoride, nitrate, sulfate, phosphate, hexafluorophosphate and dicyanamide.
- 15.-20. (Cancelled)